

A NOVEL CARBANIONIC CYCLODIMERIZATION OF DIALLENYL  
SULFONES TO 2,6-DITHIAADAMANTANE DERIVATIVES<sup>1</sup>

Samuel Braverman\*, David Reisman, Milon Sprecher\*

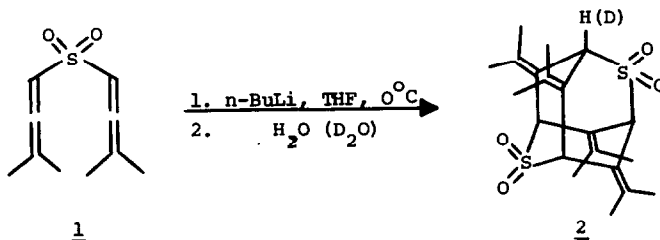
Department of Chemistry, Bar-Ilan University, Ramat-Gan, Israel

Dov Rabinovich\* and Felix Frolow

Department of Structural Chemistry, The Weizmann Institute of Science,  
Rehovot, Israel

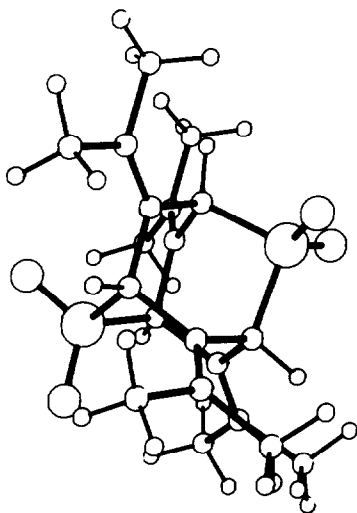
The thermal cyclodimerization of allenes to 1,2-dimethylenecyclobutanes has been of long-standing interest because of its synthetic utility and of its mechanistic aspects. Consequently, a vast amount of work has been performed on monoallenes<sup>2</sup>. In contrast, the study of diallenic systems has received relatively little attention in the past<sup>3</sup>. Recently, we have reported that bis- $\gamma,\gamma$ -dimethylalleny sulfone (1) undergoes a facile and quantitative thermal cyclization to 3-isopropenyl-4-isopropylthiophene-1,1-dioxide<sup>3a</sup> and a spontaneous electrophilic fragmentation-cyclization to  $\alpha,\beta$ -unsaturated  $\gamma$ -sultines<sup>3e</sup>.

In continuation, we have also studied the behaviour of the diallenyl sulfone 1 under basic conditions. Surprisingly, we have found that treatment of a solution of 1 in tetrahydrofuran or ether with *n*-butyllithium at 0° results in a novel and fast dimerization-cyclization of the diallenyl sulfone to give the 2,6-dithiaadamantane derivative 2, mp, 260° (d); <sup>1</sup>H nmr (CDCl<sub>3</sub>, 100 MHz):  $\delta$  1.97 (s, 24H), 5.20 (s, 4H); <sup>13</sup>C nmr (CDCl<sub>3</sub>, 20 MHz):  $\delta$  21.68 (Me), 61.20 (CHSO<sub>2</sub>), 119.38 & 137.10 (C=C); ir (CHCl<sub>3</sub>), 1100 (s), 1300 (s), 1645 (w) cm<sup>-1</sup>; ms: M<sup>+</sup> absent, m/e 332 (M<sup>+</sup>-SO<sub>2</sub>), 268 (100%), 225; Calcd. for C<sub>20</sub>H<sub>28</sub>S<sub>2</sub>O<sub>4</sub>: C, 60.6; H, 7.1; S, 16.2. Found: C, 60.4; H, 7.0; S, 16.2; (yield 60%).



Although the spectral evidence was in accord with the assigned structure of 2, unequivocal structure proof was obtained by X-ray crystallographic analysis (Figure 1).

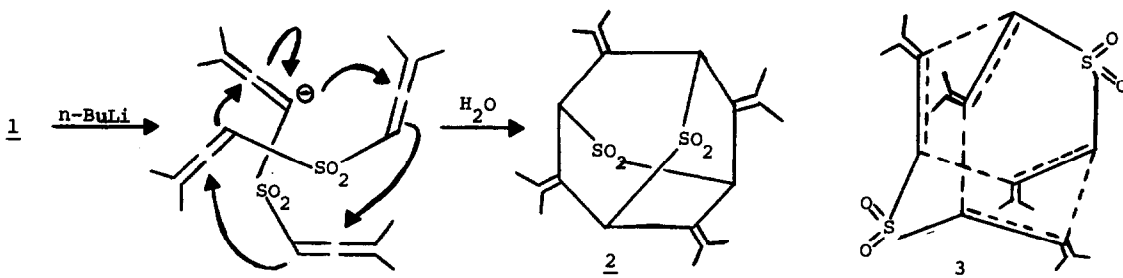
The cyclodimerization of 1 is of considerable interest on several accounts. Since diallenic sulfones can be obtained conveniently by a double [2,3]-sigmatropic rearrangement of propargylic sulfoxylates<sup>3a</sup>, the conversion of 1 to 2, represents a facile synthesis of a dithiaadamantane system. The extension of this reaction to other bridged diallenic



**Figure 1.** A structural view of 2 as determined by X-ray analysis. Crystal data: Monoclinic,  $P2_1/c$ ,  $a = 14.130$ ,  $b = 9.327$ ,  $c = 16.366 \text{ \AA}$ ,  $\beta = 106.35^\circ$ ,  $Z = 4$ . A data set of 4781 reflections was measured on a CAD3 diffractometer using Mo- $K_\alpha$  radiation. The structure was solved by direct methods and refined anisotropically by least squares using 2984 observed reflections ( $3\sigma$ ). Final discrepancy indices are  $R = 0.076$ ,  $R_w = 0.062$ . Averaged bond distances: S-O 1.432(2), S-C 1.792(3), C=C 1.329(12), C-C 1.513(11)  $\text{ \AA}$ . Full details will be published elsewhere.

systems is currently under investigation in our laboratory. To the best of our knowledge, this appears to be the first reported example of a base-catalyzed cyclodimerization of an allenic system. The dimerization of 1 may be explained by a novel mechanism (Scheme I) in which an allenic  $\alpha$ -sulfonyl carbanion initiates a process named by us "carbanion walk" or "carbanion tour" the path of which is illustrated by a dashed line in 3, with no claim for concertedness.

**Scheme I**



**REFERENCES AND NOTES**

1. Presented at the Second IUPAC Symposium on Organic Synthesis, Jerusalem-Haifa, Israel, September 10-15, 1978. See Abstracts, p.72.
2. For reviews on this subject see: (a) J.D. Roberts and C.M. Sharts, *Org. React.*, **12**, 1 (1962); (b) D.R. Taylor, *Chem. Rev.*, **67**, 317 (1967); (c) J.E. Baldwin and R.H. Fleming, *Fortschr. Chem. Forsch.*, **15**, 281 (1970); (d) T. Okamoto, *Bull. Inst. Chem. Res.*, Kyoto Univ., **50**, 450 (1972).
3. (a) S. Braverman and D. Segev, *J. Amer. Chem. Soc.*, **96**, 1245 (1974), and references cited therein; (b) H. Hauptmann, *Tetrahedron Letters*, 3589 (1974); (c) P.J. Garratt and S.B. Neoh, *J. Amer. Chem. Soc.*, **97**, 3255 (1975); (d) S. Braverman, Y. Duar and D. Segev, *Tetrahedron Letters*, 3181 (1976); (e) S. Braverman and D. Reisman, *J. Amer. Chem. Soc.*, **99**, 605 (1977); *Tetrahedron Letters*, 1753 (1977); (f) P. Blickle and H. Hopf, *Tetrahedron Letters*, 449 (1978); (g) Y.S.P. Cheng, E. Dominguez, P.J. Garratt and S.B. Neoh, *ibid.*, 691 (1978); (h) S. Braverman and Y. Duar, *ibid.*, 1493 (1978).

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